



U7845

Reg. No.: .....



Name:.....

**University of Kerala**  
 First Semester Degree Examination, November 2024  
 Four Year Under Graduate Programme  
 Discipline Specific Core Course  
**Biochemistry**  
**UK1DSCBCH101- FOUNDATIONS OF BIOCHEMISTRY**  
**Academic Level: 100 - 199**

Time: 1½ Hours

Max.Marks:42

**Part A.**  
**Answer All Questions, Objective Type. 1 Mark Each.**  
**(Cognitive Level: Remember/Understand)**  
**6 Marks. Time: 6 Minutes**

Qn No.	Question	Cognitive level	Course outcome (CO)
1	Which cell organelle is responsible for translation.	Remember	CO1
2	Name a virus with DNA as the genetic material.	Remember	CO3
3	Define a chemical bond.	Remember	CO2
4	Which type of bond is present between complementary bases on DNA?	Remember	CO3
5	What do you mean by a standard solution?	Understand	CO4
6	Define pH.	Remember	CO2

**Part B.**  
**Answer All Questions, Short Answer. 2 Marks Each.**  
**(Cognitive Level: Understand/Apply)**  
**8 Marks. Time: 24 Minutes**

Qn No.	Question	Cognitive level	Course outcome (CO)
7	Explain the functions of a light microscope.	Understand	CO4
8	Give a difference between an animal cell and a plant cell.	Understand	CO1
9	What is the pOH of a solution with an H <sup>+</sup> concentration of 0.001M?	Apply	CO2
10	State central dogma of genetics.	Understand	CO3

**Part C.**

**Answer all 4 Questions, choosing among options within each question.  
Long Answer. 7 marks each. (Cognitive Level: Apply/Analyse/Evaluate/Create)  
28 Marks. Time: 60 Minutes**

<b>Qn No.</b>	<b>Question</b>	<b>Cognitive level</b>	<b>Course outcome</b>
11	A) Describe the functions of subcellular organelles present in eukaryotes. OR	Understand	CO-1
	B) Compare and contrast the structural aspects of prokaryotic and eukaryotic cell.	Understand	CO-1
12	A) Define an acid and a base according to the Bronsted-Lowry theory. Calculate the pH and pOH of a solution with a hydrogen ion concentration of $4.3 \times 10^{-5}$ M. State whether the solution is acidic, basic or neutral. OR	Apply	CO-2
	B) Detail the properties of ester, hydrogen, ionic and van der Waals bonds and give examples of biomolecules with these bonds.	Apply	CO-2
13	A) Give a brief description on the different types of glassware used in biochemistry student's laboratory. OR	Understand	CO-4
	B) Describe the types and uses of balances used in the laboratory.	Understand	CO-4
14	A) Write a note on the properties of DNA. Describe how Meselson-Stahl experiment was carried out to demonstrate the semi conservative nature of replication. OR	Understand	CO-3
	B) Describe the transformation of R-strain pneumococcus to S-strain observed in Griffith's Experiment. How did Hershey-Chase experiment demonstrate DNA's role in genetic inheritance?	Understand	CO-3